

Chem 243 Cumulative Exam. R. Corn. Spring 2006.

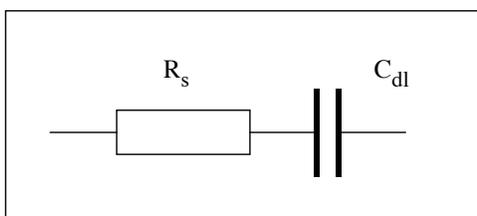
1) Electrochemical Impedance Spectroscopy (40 points)

For the following two electrochemical circuits, please

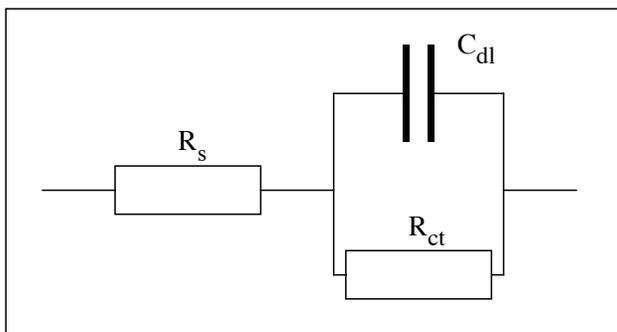
- i) derive equations for the complex impedance $Z(\omega)$;
- ii) derive equations for the Real and Imaginary parts of $Z(\omega)$;
- iii) sketch the Nyquist Impedance plot you would expect to see from this cell.

(Hint: $Z(\omega) = 1/j\omega C$ for a Capacitor)

a) Ideal Polarized Electrode:



b) Randles Cell:



2) Classical Complex Susceptibility (30 points)

Consider the equation of motion for the position $x(t)$ of a charged particle in a damped harmonic oscillator potential with a driving force $-eE(t)$:

$$m \frac{d^2x}{dt^2} + m\Gamma \frac{dx}{dt} + m\omega_0^2 x = -eE(t)$$

a) Using the initial conditions $x(t=0) = 0$ and $x'(t=0) = 0$, use Laplace transforms to find the transfer function $\mathbf{H}(s)$ that relates the Laplace transform $\mathbf{x}(s)$ to the Laplace transform $\mathbf{E}(s)$ (note: bold indicates the Laplace Transform; e.g. $x(t)$ vs. $\mathbf{x}(s)$):

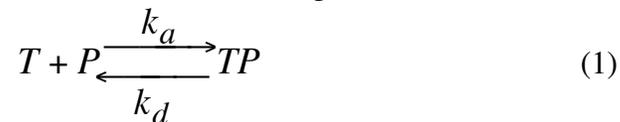
$$\mathbf{x}(s) = \mathbf{H}(s)\mathbf{E}(s)$$

b) Let $E(t) = E_0 \cos(\omega t)$ and then find the transfer function $H(\omega)$. (Hint: for this driving field, s goes to $j\omega$). Please sketch this function H as a function of ω .

c) How is this function $H(\omega)$ related to the classical complex susceptibility $\chi(\omega)$?

3) Langmuir Adsorption and Desorption Kinetics (30 points)

The vast majority of surface bioaffinity measurements utilize the specific adsorption of target biomolecules (T) from solution onto a surface that has been chemically modified with probe biomolecules (P). If the target and probe interact in a simple 1:1 ratio, then in the absence of bulk transport the surface reaction can be represented in the form:



where TP is the surface bound target-probe complex. Both P and TP are surface species, and in the Langmuir approximation their surface concentrations Γ_P and Γ_{TP} are linked to the total concentration of surface sites Γ_{tot} by eq 2:

$$\Gamma_P + \Gamma_{TP} = \Gamma_{tot} \quad (2)$$

A) If we define θ as the fraction of occupied surface sites, $\theta = \Gamma_{TP} / \Gamma_{tot}$, write down the differential equation for the time evolution of θ ($d\theta/dt = \dots$) using k_a and k_d and the bulk concentration $[T]$.

B) Find the steady state equilibrium surface coverage θ_{eq} , which is obtained in the steady state approximation ($d\theta/dt = 0$). This equation defines the Langmuir adsorption coefficient $K_{ads} = k_a/k_d$. What is θ_{eq} at a bulk concentration equal to $1/K_{ads}$?

C) Solve the differential equation in part (A) to determine the time dependence of the fractional surface coverage, $\theta(t)$, for the case where $\theta = 0$ at $t = 0$ and the bulk concentration $[T] = T_0$. Hint: you can use Laplace transform methods to solve the differential equation.